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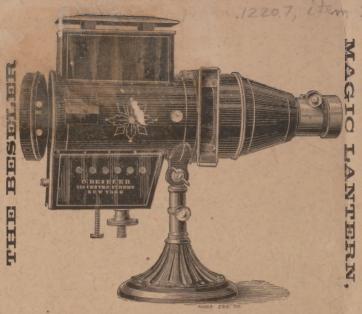
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|----|---|-----|--------------|----|----|----|----|----|------|----|------|----|------|------|----|-----|
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| | | | 60 | | | | | | | | | | | | 16 | |
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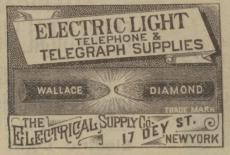
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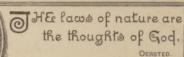
THE YOUNG CHEMIST.

QUALITATIVE CHEMICAL ANALYSIS,

QUANTITATIVE CHEMICAL ANALYSIS.

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→* CALENDAR.*

| | | | 1 | | | | | | - | | 15.1 | .21 | | | |
|----------|------|--------|---------|--------|---------|--------|---------|--|--------|--------|------|--------|--------|-----|-------|
| D | 2 | Monday | Tuesday | Wednes | Thursd. | 3 | Saturd. | 1885 | 23 | Monday | day | Wednes | Thursd | 3 | d. |
| 88 | de | nd | 380 | di | 17 | Friday | 197 | 00 | Sunday | na | 350 | 9 | 12 | do | 277 |
| 00 | 27.3 | 10 | 246 | 2 | 123 | 7. | 727 | 00 | 367. | 10 | 77 | 20 | 1/2 | 2 | at |
| Annual | S | 10 | | - | - | | | - Personal P | | - | 17 | - | - | 1 | 2 |
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| | 4 | | 6 | | | | IO | | 5 | 6 | | | | IO | |
| | | | | | 15 | | | | | | | | 16 | | |
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| Eak | | | | | *** | · · · | 12.0 | Aug. | | | | | | | |
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| | 24 | | | 25 | 20 | | | | | | | | | | 29 |
| Mar. | 7 | | 3 | 4 | | 6 | | Sept. | | 34 | | | | 4 | 5 |
| ATACKE . | | | | | | | 14 | oche. | 6 | | 8 | | 10 | | |
| | | | | | | | 21 | | | | | | 17 | | |
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| May | | | | | | I | 2 | | | | | | | | 55.0 |
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| | 28 | 29 | 130 | 555 | 500 | | | | 27 | 28 | 129 | 30 | 31 | 556 | 5.7.5 |

of all the year were playing folidays To sport would be as tedious as to work.

FIRST PART-HENRY IV.

EDITORIAL GREETING.

The third annual issue of

THE LABORATORY HANDBOOK

is now offered for the favorable consideration of the scientific world. Many of our friends have found the previous numbers serviceable; they will discover in this one several new features that we trust will merit their approval.

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THE EDITOR.

ASTRONOMICAL NOTES FOR 1885.

ECLIPSES.

In the year : s there will be four eclipses, two of the sun and two of the moon.

1. An annular celipse of the sun on March P, visible at Washington as a partial eclipse.

CR. CMSTANCES OF THE E THESE.

| | н. м. | |
|-------------------------|--------|-------|
| Letipse legins | 10.00 | A. M. |
| Central eclipse begins | 11 11 | . 6 |
| Central eclipse at noon | . г об | P. M. |
| Central eclipse ends | I 43 | . ** |
| Eclipse ends | . 3 06 | 6.6 |

II. A partial eclipse of the moon on March 10, invisible at Washington; visible in the Western Partic Ocean, Asia and the eastern portion of Europe and Africa. 111. A total eclipse of the sun on September 8, invisible at Washington, but wisible in the South Pacific Ocean.

IV. A partial celipse of the moon on September 23-24, visible at Washington; also on the Atlantic Ocean, North and South America, and the Pacific Ocean.

TIMES OF THE PHASES.

| 1 | I. M. | |
|-----------------------|-------|-------|
| Moon enters penumbra | 1 52 | P. M. |
| Moon enters shadow | (10 I | A. M. |
| Middle of the eclipse | 2 40 | 44 |
| Moon leaves shadow | 4 14 | 6.0 |
| Moon leaves penumbra | - + + | 4.6 |

First Month.

| | JANUARY | | | | | | | | | | | |
|---------------|---------|----|----|----|----|----|--|--|--|--|--|--|
| s m t w t f s | | | | | | | | | | | | |
| | | | | 1 | 2 | 3 | | | | | | |
| 4 | 5 | 6 | 7 | _ | 9 | 10 | | | | | | |
| 11 | | | | | 16 | | | | | | | |
| | 19 | | | | | | | | | | | |
| 25 | 26 | 27 | 28 | 29 | 30 | 31 | | | | | | |
| | | | | | | | | | | | | |

31 Days.

1885.

| | FEBRUARY | | | | | | | | | |
|---------------|----------|----|----|----------|----|----|---|--|--|--|
| | N | M | T | w | E | 15 | 3 | | | |
| | 1 | 2 | 3 | 4 | 5 | | 7 | | | |
| Second Month. | 8 | 16 | 10 | | 12 | | | | | |
| | 13 | | | 18 25 | | | | | | |
| | | | | | 20 | | | | | |
| | | | | | | | | | | |

28 Days.

1885.

| | →* MARCH * | | | | | | | | | | |
|----|------------|----|----|----|----|----|--|--|--|--|--|
| 3 | 1Ht | E | aw | T | F | 3 | | | | | |
| 1 | 2 | 3 | 4 | 5 | 6 | 7 | | | | | |
| 8 | | | 11 | | | | | | | | |
| | | | 18 | | | | | | | | |
| | | | 25 | 26 | 27 | 28 | | | | | |
| 29 | 30 | 31 | | | | | | | | | |
| | | | | | | | | | | | |

Third Month.

31 Days.

1885.

| Fourth | Month. |
|--------|--------|

| →# APRIL #← | | | | | | | | | | | |
|---------------|---------------|---------------|---------------|---------------|---------------------|----------------|--|--|--|--|--|
| 3 | M | 2 | w | e | F | 3 | | | | | |
| 5 12 19 | 6 13 20 | 7 14 21 | 8 15 22 | 9 16 23 | 3 10 17 24 | 11 18 25 | | | | | |
| | | | | | | | | | | | |

30 Days.

1885.

| | →* MAY * | | | | | | | | | | | |
|---------------|----------|----|----|----|----|----|--|--|--|--|--|--|
| s m t w t f s | | | | | | | | | | | | |
| | | | | | 1 | 2 | | | | | | |
| 3 | 4 | | | 7 | | | | | | | | |
| | 11 | | | | | | | | | | | |
| 17 | 18 | 19 | 20 | 21 | 22 | 23 | | | | | | |
| 24 | 25 | 26 | 27 | 28 | 29 | 30 | | | | | | |
| 31 | | | | | | | | | | | | |

Fifth Month.

31 Days.

1885.

| | | 4) | * J | UN | E米 | ~ | |
|-------------|---|----|-----|----------|---------------------|----|----|
| | 3 | M | T | w | T | F | 30 |
| ixth Month. | | 22 | 16 | 17 24 | 4 11 18 25 | 12 | 20 |
| | | | | | | | |

30 Days.

1885.

Seventh Month.

| | →# JULY **← | | | | | |
|----|-------------|----|-----|----|----|---|
| 5 | PA | E | TT! | T | 15 | 3 |
| | | | | | 3 | |
| | | | 8 | | | |
| | | | 15 | | | |
| | | | 22 | | | |
| 26 | 27 | 28 | 29 | 30 | 31 | |
| | | | | | | |

31 Days.

1885.

| MEN | IOR. | AN | DA. |
|-----|------|----|-----|
|-----|------|----|-----|

| AUGUST | | | | | | | |
|--------|----|----|----|-------|----|-----|--|
| B | M | T | æ | T | F | S | |
| 2 | 3 | | 5 | 6 | 7 | 1 8 | |
| 16 | 10 | 11 | | 13 20 | | | |
| 23 | 24 | 25 | 26 | 27 | 28 | 29 | |

Eighth Month.

31 Days.

1885.

| | SEPTEMBER | | | | | |
|----|-----------|---|---------------|----------------|---------------------|----------|
| 3 | H | T | E | T | F | = |
| 20 | 21 | | 9 16 23 | 10 17 24 | 4 11 18 25 | 12 19 |
| | | | | | | |

30 Days.

Ninth Month.

1885.

| MEMORANDA. | | | | |
|------------|----|----|-----|-----|
| | MO | DA | NIT | 7 1 |

| OCTOBER | | | | | | |
|---------|----|----|----|----|----|----|
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| 4 | 5 | | 7 | | _ | 10 |
| 11 | | | | 15 | | |
| | | | | 22 | | |
| 25 | 26 | 27 | 28 | 29 | 30 | 31 |
| | | | | | | |

Tenth Month.

31 Days.

1885.

Eleventh Month.

| | N | OV | EM | BE | R | |
|----|-------|----|----|----|----|----|
| 3 | M | T | E | T | F | 3 |
| 1 | 2 | | - | 5 | - | 7 |
| 8 | 1 | | 11 | | | |
| | 16 | | | | | |
| | 23 | 24 | 25 | 26 | 27 | 28 |
| 29 | 30 | | | | | |
| | , , . | | | | | |

30 Days.

1885.

| DECEMBER | | | | | | |
|--------------|------------------------------------|----------|----------------|----|----|--|
| SI | HE | E | E | F | S | |
| 13 1 20 2 | . 1 7 8 4 15 1 22 8 29 | 16 23 | 10 17 24 | 18 | 12 | |

Twelfth Month.

31 Days.

1885.

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"I think the work will prove exceedingly valuable to that large class of young persons who would like to pursue chemical studies and are not permitted to have the guidance of a competent teacher."

From Professor A. B. AUBERT,

Maine State College, Orono, Me.

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"The explanatory notes are full and make the study of quantitative determinations easy to students and time saving to the teacher."

From Professor B. F. MORLEY,

Pennsylvania Military Academy, Chester, Pa.

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From Professor M. B. HARDIN,

Virginia Military Institute, Lexington, Va.

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From Professor H. CARRINGTON BOLTON, Ph.D.,

Trinity College, Hartford, Conn.

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From Professor H. C. COON, M. D.,

Alfred University, Alfred Centre, N. Y.

"I think the works that I have used, Appleton's Young Chemist and Appleton's Qualitative Analysis, are very excellent guides to students in their laboratory work. They are works in the right direction. I like them very much."

From Professor EDWARD HART, Ph.D.,

Lafayette College, Easton, Pa.

"Appleton's Quantitative Analysis seems to be an excellent work."

From Professor G. E. PATRICK,

Kansas State University, Lawrence, Kansas.

"Appleton's Quantitative Analysis is just the thing for beginners in that branch of chemistry. It is explicit without being tiresome, and it will save the instructor much time and many words."

From Professor T. C. VAN NUYS,

Indiana State University, Bloomington, Ind.

"As far as I have been able to examine Appleton's Quantitative Analysis, I am very favorably impressed with it."

From Professor A. H. SABIN, M. S.,

University of Vermont, Burlington, Vt.

"I have received Appleton's Quantitative Analysis, and after using it, can say that I like it very much. It is exactly what is needed: a cheap and thorough book on the subject for students who are not taking a very extended course.

"There is not a laboratory in the country where this book would not be

valuable for teaching purposes."

From Professor JAMES A. DODGE,

University of Minnesota, Minneapolis, Minn.

"I have examined the present edition of Appleton's Quantitative Analysis, and consider it an excellent book, very well suited to our purpose. I think it probable that we shall shortly make this a regular required hand-book for beginners in this branch."

From Professor R. W. JONES, M. A.,

University of Mississippi, Oxford, (Lafayette Co.) Miss.

"I commenced last session the use of Qualitative Analysis, and like it. I am using it with the beginners this session."

From Professor CHARLES R. FLETCHER, S. B.,

Boston University, Boston, Mass.

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From Professor J. C. FALES,

Centre College, Danville, Ky.

"If Professor Appleton's Quantitative Analysis proves as satisfactory as his Qualitative Analysis in the hands of beginners, it will be a valuable help."

From Prof. P. de P. RICKETTS, E. M., Ph. D.,

Columbia College School of Mines, New York City.

"I have carefully examined Appleton's Quantitative Analysis and consider it one of the best little works I have ever seen on the subject.

"I admire particularly the clear and explicit way of stating the reasons for the adoption of processes of analysis and use of various reagents."

From Professor CHAS. A. GOESSMANN, Ph. D.,

Massachusetts Agricultural College, Amherst, Mass.

⁶ I have examined Appleton's Quantitative Analysis with much pleasure, and consider it a very valuable addition to our chemical literature for the instruction of beginners in practical analytical chemistry.

"The technical execution of the work is not less praiseworthy than the judicious methods adapted to meet the wants of the student when entering the field of quantitative analysis and to assist materially the teacher in his task of intelligent instruction."

From Professor ALONZO COLLIN,

University of Nebraska, Lincoln, Neb.

"I have been using Appleton's Quantitative Analysis this session, and can say that I am much pleased with it."

From Professor GHARLES A. SCHAEFFER, Ph. D., Cornell University, Ithaca, N. Y.

"I had postponed looking over Appleton's Quantitative Analysis until I could examine it thoroughly. This I have done, and will venture to express my hearty approval of the book. For the purpose for which it is intended, that is, as a guide for beginners, it is certainly well written and thoroughly, and I shall be very glad to recommend it to all students commencing Quantitative Analysis."

From Professor H. B. CORNWALL,

John C. Green School of Science, Princeton, N. J.

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Columbus, Ohio.

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From Professor LA ROY F. GRIFFIN, A. M.,

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|------------------------------|------|-------|
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| Over \$30 and not exceeding | 40 | 1 25 |
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|--|--|
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| On application for design (7 years) | |
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TABLE BASED ON THE LATEST REVISION OF

ATOMIC AND MOLECULAR WEIGHTS.

(Derived from Professor F. W. Clarke's figures.)

| | | 1 | | | f |
|--|--------------------|-------------|---|------------|-------------|
| AMES OR FORMULAS. | WEIGHTS. | Weights. | NAMES OR FORMULAS. | WEIGHTS. | Weight |
| Muminum, Al | 27.0000 | 27. | Cobalt, Co | 58.8870 | 58.0 |
| Al ₂ O ₃ | 101.9079 | 101.9 | Copper, Cu | 63.1730 | 63.2 |
| $Al_2(SO_4)_3 - (NH_4)_2$ | | | CuO | 79.1363 | 79.1 |
| SO ₄ 24H ₂ O | 904.5280 | 904.5 | CuSO ₄ 5H ₂ O | 248.8267 | 248.8 |
| Antimony, Sb | 119.9550 | 120. | Didymium, D | 144.5730 | 144 (|
| Sb_2O_4 Sb_2S_3 | 303.7632 | 303.8 | Erbium, E | 165.8010 | 165.0 |
| K Sb O H ₂ O ₄ C ₄ H ₂ | 335.0020 | 333.9 | Fluorine, Fl | 18.9840 | |
| O2 + 1/2 H2O | 331.5931 | 331.6 | | | 19. |
| Arsenic, As | 74.9185 | 74.9 | Gallium, Ga | 68.8540 | 68. |
| As ₂ O ₃ | 197.7259 | 197.7 | Glucinum, G | 9 0850 | 9. |
| As ₂ S ₃ | 245.7880 | 245.8 | Gold, Au | 196.1550 | 196.: |
| NH ₄ Mg As O ₄ 12 | 40 | 206.2 | Hydrogen, H | 1,0000 | 1. |
| $H_2() \dots Mg_2As_2O_7 \dots$ | 396.3108 | 396.3 | H ₂ O | 17.9633 | 18. |
| Barium, Ba | 136.7630 | 136.8 | Indium, In | 113 3980 | 113 |
| BaCl ₂ + 2H ₂ O | 243.4296 | 243.4 | | | |
| BaSO ₄ | 232.6002 | 232.6 | Iodine, I | 126.5570 | 126. |
| Bismuth, Bi | 237.5230 | 207.5 | AgI | 165.5760 | 165. |
| Bi ₂ O ₃ | 462.9359 | 462.9 | | 234.2320 | 234. |
| BiONO ₃ | 285.3972 | . 285.4 | Iridium, Ir | 192.6510 | 192. |
| Boron, Bo | 10.9410 | 10.9 | Iron, Fe | 55.9130 | 55.1 |
| Brømine, Br | 79.7680 | 79.8 | Fe ₂ O ₃ | 159.7159 | 159. |
| KBr | 118.7870 | 118.8 | FeSO ₄ +7H ₂ O | 277 - 4933 | 277. |
| AgBr | 187.4430 | 187.4 | Fe SO ₄ + (NH ₄) ₂ SO ₄ 6H ₂ O | | |
| Cadmium, Cd | 111.8350 | 111.8 | Fe ₂ (SO ₄) ₃ (NH ₄) ₂ | 391.4092 | 391. |
| CdO | 127.7983 | 127.8 | SO ₄ 24H ₂ O | 962.3360 | 962. |
| | 143.8190 | 143.8 | | | |
| Cæsium, Cs | 132.5830 | 132.6 | Lanthanum, La | 138.5260 | 138. |
| Calcium, Ca | 39.9900 | 40. | Lead, Pb | 206.4710 | 206. |
| CaO | 55.9533 99.8535 | 56. 99.9 | PbO | 222-4343 | 222. |
| CaSO ₄ | 135.8272 | 135 8 | PbS | 238.4550 | 238. |
| Carbon, C | 11.9736 | 12. | PbSO ₄ | 330.2920 | 302. |
| CO | 27.9369 | 27.9 | | 33 | 3 |
| CO ₂ | 43.9002 | 43.9 | Lithium, Li | 7.0073 | 7 - |
| Cerium, Ce | 140.4240 | 140.4 | 35 | | |
| Chlorine, Cl | 35.3700 | 35.4 | Magnesium, Mg | 39.9223 | 24 . |
| AgCl | 143.0450 | 143. | MgSO ₄ + 7H ₂ O | 245.5393 | 245 |
| NaCl | 58.3680 | 58.4 | Mg.P.O | 221.5771 | 221. |
| Chromium, Cr | 52.0090 | 52. | | | |
| Cr ₀ () ₃ | 151.9079 | 151.9 | Manganese, Mn | 53.0000 | 53 - |
| $K_2Cr_2O_7K_2SO_4 - Cr_2(SO_4)_3$ | 293.7991 | 293.8 | MnO | 69.8693 | 69.1 85. |
| 24H.O | | | MnO2 | 85.8326 | 05. |

TABLE BASED ON THE LATEST REVISION OF

ATOMIC AND MOLECULAR WEIGHTS.

(Continued.)

| | | | | 1 | 1 |
|--|----------|------------------------|--|-----------|------------------------|
| NAMES OR FORMULAS. | WEIGHTS. | Approxim'e Weights. | NAMES OR FORMULAS. | WEIGHTS. | Approxim's Weights. |
| Mercury, Hg | 199.7120 | 199.7 | Rubidium, Rb | 85.2510 | 85.3 |
| Hg('l2 | 270.4520 | 270.5 | Ruthenium, Ru | 104.2170 | |
| HgS | 231.6960 | 231.7 | | | 104.2 |
| Molybdenum, Mo | 95.5270 | 95.5 | Scandium, Sc | 43.9800 | 44 - |
| | | | Selenium, Se | 78.7970 | 78.8 |
| Nickel, Ni | 57 9280 | 57.9 | Silicon, Si | 28.1950 | 28.2 |
| NiO Ni SO ₄ (NH ₄) ₂ | 73.8913 | 73.9 | SiO ₂ | 60 1216 | 60.1 |
| SO ₄ 6H ₂ O | 393.4242 | 393.4 | Silver, Ag | 107.6750 | 107.7 |
| Niobium, Nb | 93.8120 | 93.8 | AgCl | 143.0450 | 143. |
| | | 1 | AgNO ₃ | 169.5859 | 169.6 |
| Nitrogen, N | 14 0210 | 14. | Sodium, Na | 22.9980 | 23. |
| HNO3 | 62.9109 | 62.9 | NaCl | 58.3680 | 58.4 |
| KNO ₈ | 100.9299 | 100.9 | Na ₂ CO ₃ | 105.8595 | 105.9 |
| NH., | 17.0210 | 17. | Na ₂ O | 61.9593 | 62. |
| NH ₄ | 18.0210 | 18. | Strontium, Sr | 87.3740 | 87.4 |
| NH ₄ Cl NH ₄ OH | 53.3910 | 53.4 | SrSO ₄ | 183.2112 | 183.2 |
| (NH ₄) PtCl ₆ | 442.0770 | 442.7 | Sulphur, S | 31.9840 | 32. |
| Osmium, Os | 198.4940 | 198.5 | SO ₂ | 63.9106 | 63.9 |
| | | | SO ₃ | 79.8739 | 79.9 |
| Oxygen, O | 31.9266 | 16. 31.9 | SO ₄ H ₂ SO ₄ | 95.8372 | 95.8 97.8 |
| 03 | 47.8844 | 47.9 | | | |
| 04 | 63.8532 | 63.9 | Tantalum, Ta | 182,1440 | 182.1 |
| ()3, | 79.8105 | 79.8 | Tellurium, Te | 127.9600 | 128. |
| 0, | 95.7798 | 95.8 | Thallium, Tl | 203 7150 | 203.7 |
| 0, | 111.7431 | 111.7 | Thorium, Th | 233.4140 | 233.4 |
| 0, | 143.6097 | 143.7 | Tin, Sn | 117.6980 | 117.7 |
| Palladium, Pd | 105.7370 | 707 - | SnCl ₂ +2H ₂ O | 224.3646 | 224.4 |
| | | 105.7 | SnO ₂ | 149 6246 | 149.6 |
| Phosphorus, P | 30.9580 | 31. | *Titanium, Ti | 47 - 9997 | 48. |
| P ₂ O ₅ | 141.7325 | 221.6 | Tungsten, W | 183.6100 | 183.6 |
| | 221.5771 | 221.0 | Uranium, U | 238.4820 | 238.5 |
| Platinum, Pt | 194.4150 | 194-4 | | | |
| K ₂ PtCl ₃ | 434.6730 | 484.7 | Vanadium, Va | 51.2560 | 51.3 |
| | | 444.7 | Ytterbium, Yb | 172.7010 | 172.8 |
| Potassium, K | 39.0190 | 39 - | Yttrium, Y | 89.8160 | 89.8 |
| K.O | 74-3500 | 74 4 | Zinc, Zn | 64.9045 | 64.9 |
| K ₂ PtCl ₆ | 484.6730 | 484.7 | ZnO | 80.8678 | 80.9 |
| K ₂ SO ₄ | 173.8752 | 173.9 | ZnSO ₄ 7H ₂ O | 286.4848 | 286.5 |
| Rhodium, Rh | | 104.1 | Zirconium, Zr | 89.3670 | 80.4 |
| AND ALLER TO THE PARTY OF THE P | 104.0750 | 104.1 | 27110-7111111111, 271 | 39.3.70 | 09.4 |

^{*}Thorpe, T. E., Chemical News, 48; 251.

TABLES OF WEIGHTS AND MEASURES.

ENGLISH WEIGHTS.

| | | | TROY | WEIGH | T. | | |
|----------|---|----------|---------------|---------------|---------|----------|-----------------|
| Po | nund. O | unces. | | eights. | | rains. | French Grammes |
| | | | 2 | 10 | 5 | | = 373-2419 |
| | | | | | | | = 31.1035 |
| | | | | I | | 24 1 | = 1.5552 |
| B10.7007 | *************************************** | | APOTHEC | ADTES! W | FIGHT | | |
| | lb. | 3 | 3 | 8 | LIGHT | gr. | |
| Po | | | rachms. | Scruples | . Gra | | French Grammes |
| | | | 96 | | | | = 373.2419 |
| | | | 8 | | | | = 31.1035 |
| | | | I | 3 | | 60 = | 3 8579 |
| | | | | I | | | 1.2959 |
| | | | | | | T : | = .0648 |
| | | | AVOIRDU | POIS WE | IGHT. | | |
| Po | nund. | Ounces. | Dra | chms. | Gi | rains. | French Gramme: |
| | I | . 16 | 2 | 56 | 70 | 100 2 | = 453.5926 |
| | | I | | 26 | 4 | 37.5 | 28.3495 |
| | | | | X | | 27.343 | 1.7718 |
| | | | | | | | |
| | | | METRIC | C MEAS | URES. | | |
| | | | MEASURE | S OF LE | ENGTH. | | |
| I | Millimetre | = 0.0 | ooi of a me | re. | | | |
| I | Centimetre | | oro of a me | | | | |
| I | Decimetre | | 100 of a mel | | about 4 | inches. | |
| 1 | Metre | | 00 Metre | 2000 | 39.37 | inche | 8. |
| X | Decametre | | ooo metres. | | | | |
| | Hectometre | | ooo metres. | | 3.5.0 | , , | ** |
| I | Kilometre | | ooo metres | | | of a m | |
| I | Myriametre | 10,000.0 | opo metres | and | about o | 1 miles. | |
| - | | | MEASURI | ES OF SU | RFACE | | |
| I | Centiare | т т | Square met | re | == al | out 11 | somare vards |
| | Are | | Square 1 | | | . 5 | square yards. |
| E | | | Square met | | | out 21/2 | |
| - | | | *********** | 70 00 110 | TATALE | | |
| _ | Cubic Metre | | MEASURI | | LUME. | | |
| X | Cubic Metre | | itres, or one | | | | |
| | | | Stere. | . Ithomete. | | | |
| ****** | MA | | | CORCAT | ACITA | | |
| | | | MEASURE | | ACITY. | | |
| 1 | Litre | | cubic decir | | == 8 | bout 1 | quart. |
| | | 7 01 10 | | itiliteties) | | | |
| | | | MEASUR | ES OF W | EIGHT. | | |
| 1 | Milligramme | === | 0.001 of a | | 1.2 | abou | t 1 of a grain. |
| Z | Centigramme | 202 | 0.010 of a | | | | 6.5 |
| I | Decigramme | - | 0.100 of a | | | | |
| | Gramme | == | 1.000 G | | | abo | ut 15% grains. |
| I | Decagramme | 2002 | 10.000 grai | | | | |
| I | Hectogramme Kilo(gramme) | == 1 | 100.000 grai | | - | about a | 1 1he |
| I | Kilo(granime) | 1 | ooo.ooo grai | mics | | anout 2 | 5 105. |
| 1 | Tonneau | == 10 | 000. Ki | lo's | 2073 | about 1 | ton. |
| | | | | | | | |

ALPHABETICAL TABLE OF EQUIVALENT VALUES OF WEIGHTS AND MEASURES.

| 1 Are 100 sq. metres | 119.6 sq. yards. |
|---|----------------------|
| T Centiare = 1 sq. metre, = 100 sq. centimetres | 1550. sq. inches. |
| z Centimetre = $\frac{1}{100}$ of a metre, | .3937 inches. |
| 7 Cubic centimetre; (of dist. water, weighs 1 gm.) | .o610 cub. inches. |
| z Cubic decimetre, (same as z litre) | 1000 C. C. |
| of distilled water, weighs 1000 gms., or | 1 kilogramme. |
| 7 " in English or imperial measure | .8804 quarts. |
| in American or wine measure | 1.0567 quarts. |
| 1 Cubic foot (1728 cubic inches) | ,315.3119 cub. cent. |
| " " of water (at 62° F.) weighs | 62.3210 lbs. Av. |
| 1 Cubic inch | 16.3861 cub. cent. |
| g " of water (at 62 F.) weighs | 252.458 grains. |
| 1 " of water (at 60° F.) weighs | 252.5 grains. |
| z Cubic metre (z Stere), = z,000,000. C. C., or | 1000. litres. |
| r Fluid ounce, imperial, 28.4 C. C | 1.7329 cub. inches. |
| t " wine measure, = 29 5 C. C | 1.8047 cub. inches. |
| my imperial, of water (62° F.) weighs | 437.5 grains. |
| z " wine measure, of water (60° F.) weighs | 456.0 grains. |
| : Foot | 30.48 centimetres. |
| I Gallon, imperial, = 277.274 cubic inches | 4.5435 litres. |
| of water, weighs (62° F.) 10 lbs. or | 70,000 grains. |
| 2 Gallon, wine measure,= 231. cubic inches | 3 7852 litres. |
| 2 " of water, weighs (60° F.) 8.34 lbs. or | 58,372.2 grains. |
| z Gramme (weight of z C. C. of dist. water, 4° C.) | 15.4323 grains. |
| 1 Inch | 2.54 centimetres. |
| 1 Kilogramme (1000 grammes) | 2.2046 lbs. Av. |
| 1 Litre (see cubic decimetre). | |
| 1 Metre (1 40-mill'th of Earth's meridian) 3 ft. 3 in. 3/8 in., nearly. | 39.3708 inches. |
| ¹ Pint, wine meas., = 16 fluid oz. = of water (60° F.) 7296.5 gr | 473.148 cub. cent. |
| " imperial, = 20 fluid oz. = of water (62° F.) 8750. gr | 567.932 cub. cent. |
| I Quart, wine measure, = 32 fluid ounces | .9463 litres. |
| " imperial, 40 fluid ounces | 1.1358 litres. |
| 1 Ton Avoirdupois (2000 lbs.). | 29,166%. oz. Troy. |
| r Ienneau, - 1,000,000 gms | 1000 kilo's. |
| | |

TABLE SHOWING CORRESPONDING DEGREES ON THE SCALES OF THE FAHRENHEIT AND CENTIGRADE

THERMOMETERS.

| | | | | | | | | | | | - |
|--------|------------|----------|-------|--------|--------|---------|----------|--------|-------------|--------|---------|
| FAHR. | CENT. | FAHR. | CENT. | TAHR. | CENT. | FAHR. | CENT. | TAHR. | CENT. | FAHR. | CENT. |
| 32 | O. | 62 | 16.6 | 91.4 | 33. | 121 | 49.4 | 152 | 66.6 | 182 | |
| 33 | .5 | 62 6 | 17. | 92 | 33.3 | 122 | 50 | 152 0. | 67 | 183 | 83 8 |
| 34 | T.T | 64 | 17.7 | 93.2. | 33 8 | 123.8 | 50.5 | 153 | 67.2 | 183.2. | 84 84 4 |
| 35 | 16 | 64.4 | | 194 | 34 - 4 | 124 | 51.1 | 154.4. | 68. | 185 | 85 |
| 35.6 | 2, | 65 | 18.3 | 95 | 35. | 125 | 51.6 | 155 | 68.3 | 186 | 85.5 |
| 36 | 2.2 | 66 . | 18.8 | ,96 | 35.5 | ,125.0. | 52. | 156 | 68.8 | 186.3. | 86. |
| 37 - 4 | 3. | 65.2 | 19.4 | 96.8 | 36.1 | 126 | | 156.2. | 69. | 188 | 86 1 |
| 38 | 3.3 | 68 | 20. | 98 | 36.6 | 127.4. | 52 - 7 - | 157 | 70. | 188.0. | 87 |
| 39 | 3.8 | 60 | 20.5 | 98.6 | | 128 | | 1 | | .0. | |
| 39.2 | 4. | 69.8 | 21. | 90 | 37.2 | 129 | 53.8 | 159.8. | 70.5 | 189 | 87.2 |
| 40 | 4-4 | 70 | 21.1 | 100 | 37.7 | 129.2. | 54 · | 160 | 71.1 | 190.4. | 88 |
| 42 | 5.5 | 71.6 | 21.6 | 100.4. | 38.3 | 130 | 54.4 | 161.6. | 71.6 | 101 | 88. 8 |
| | 5.5 | | 22. | 201 | 30.3 | 131 | 55 · | 101.0. | 72. | 192 | 00 0 |
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| 43 | 6.I 6.6 | 73.4 | 22.7 | 102.2. | 39 - 4 | 132.8. | 56. I | 163.4. | 72.7 73. | 194 | 39 4 |
| 44.6 | 7 - | 74 | 23.3 | 104 | 40. | 134 | 56.6 | 164 | 73.3 | 195 | 90 5 |
| 45 | 7.2 | 75 | 23.8 | 105 | 40.5 | 134.6. | 57- | 165 | 73.8 | 195 8. | C.T. |
| 46 | 7.7 | 75.2 | 24. | 105.8. | 41. | 135 | 57.2 | 165.2. | 74 - | 190 | uI.I |
| 40.4 | | 76 | 24.4 | 100 | 41.1 | 136.4. | 57.7 | 167 | 74 - 4 | 197 | 91.6 |
| 48 | 8 8 | 78 | 25.5 | 107.6. | 42. | 137 | | 168 | 75 · 5 | 197.0. | 92.2 |
| 48.2 | 9. | 78.8 | 26. | 108 | 42.2 | 138 . | 58.8 | 168.8. | 76. | 199 | 92.7 |
| 49 | 9.4 | 79 | 26.1 | 109 | 42.7 | 138.2. | 59. | 169 | 76.1 | 199.4. | 03. |
| 50 | IO. | 80 | 26.6 | 109.4. | 43. | 139 | | 170 | 76.6 | 200 | 93.3 |
| 51.8 | 10.5 | 80.6 | 27.2 | 110 | 43.8 | 140 | 60. | 170.6. | 77 - 2 | 201.2. | 93.8 |
| 52 | II.I | 82 | 27.7 | III.2. | 44. | 141.8. | 61. | 172 | 77 - 7 | 202 | 04.4 |
| 53 | 11.6 | 82.4 | 28. | 112 | 44.4 | 142 | 61.1 | 172.4 | 78 | 203 | 05. |
| 53.6 | 12 | 83 | 28.3 | 113 | | 143 | 61.6 | 173 | 78.3 | 204 | 95.5 |
| 55 | 12.2 | 84.2 | 28.8 | 114.8. | 45.5 | 143.6. | 62.2 | 174 | 78 8 | 204.8. | 96. |
| 55.4 | 13. | 85 | 29.4 | 115 | 46.1 | 145. | 62.7 | 175 | 79 - 4 | 205 | 96 1 |
| g6 | 22.5 | 86 | | 116 | | | | 176 | | | |
| 57 | 13.3 | 87 | 30.5 | 116.6. | 46.6 | 145.4. | 63. | 177 | 80.5 | 200.0. | 97.2 |
| 57 2 | 14 | . 87 . 8 | 31. | .117 | 47.2 | 147 | 03.8, | 177.8. | 81. | 208 | 97.7 |
| 58 | 14.4 | 88 | 31.1 | 118 | 47.7 | 147.2. | | 178 | 81.1 | 208.4. | |
| , | 15. | | 31.0 | -10.4. | | 140 | 04.4 | | | 209 | 98.3 |
| 60.8 | 15.5 | 89.6 | 32. | 119 | 48.3 | 149 | 65. | 179.6. | | 210 | 2 |
| 61 | 16. | 90 | 32.2 | 120 | 48.8 | 150.8. | 65.5 | 180 | | 210.2. | 99. |
| | | 1 | 3 | | 49. | 151 | 66.I | 181.4. | | 212 | |
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SPECIFIC GRAVITIES, OF SOLID AND LIQUID ELEMENTS. (WATER=1.)

| | | 1 | |
|-------------------|-------------|-------------------|---------------|
| Lithium | -59 | Zinc | 7.10 7.20 |
| Potassium | .86 | Tin | 7.29 - 7.30 |
| Sodium | .97 | Iron | 7.79 - 7.84 |
| Chlorine (liquid) | 1.33 | Manganese | 8.or — 8.o3 |
| Calcium | 1.58 | Cobalt | 8 49 - 8.51 |
| Magnesium | 1.70 - 1.74 | Nickel | 8.60 8.82 |
| Phosphorus | 1.83 1.96 | Cadmium | 8.45 - 8.69 |
| Sulphur | 1.98 — 2.07 | Molybdenum | 8.62 - 8.64 |
| Glucinum | 2.10 | Copper | 8.93 — 8.95 |
| Carbon | 2.27 - 3.52 | Bismuth | 9.78 — 9.80 |
| Silicon | 2.49 | Silver | 10.40 - 10.57 |
| Aluminum | 2.50 — 2.67 | Rhodium | 11.00 - 11.20 |
| Strontium | 2.54 | Lead | 11.33 - 11.39 |
| Bromine (liquid) | 2.99 — 3.19 | Palladium | ri.80 |
| Selenium | 4.28 — 4.80 | Mercury (liquid.) | 13.60 |
| Iodine | 4.95 | Tungsten | 17.20 18.30 |
| Arsenic | 5.63 - 5.67 | Uranium | 18.40 |
| Tellurium | 6.18 - 6.24 | Gold | 19.26 19.34 |
| Antimony | 6.72 | Platinum | 21.50 |
| Chromium | 7.01 | Iridium | 21:80 |
| | | 11 | - : |

SIMPLE FORMULAS FOR CALCULATING AREAS, SURFACES AND VOLUMES.

π - 3.1416.

PLANE AREAS.

| Triangle, (altitude, a; base, b) | Area | - | | | | | 1/2 | ab. |
|---|------|---|-------|----------------|----|----|-------|----------------|
| Circle, (radius, R; diameter, D) | 66 | = | π | \mathbb{R}^2 | or | 34 | π | \mathbb{D}^2 |
| Ellipse, (semi-axes, a and δ) | 66 | - | | | | | π | ab. |
| CUIDEACES OF SOLIDS | | | | | | | | |

Sphere, (radius, R; diameter, D)....Surface = $4 \pi R^2$ or πD^2 .

Cylinder, (radius, R; height, h)..... $(2\pi R)h = 2\pi R(R \cdot h)$.

VOLUMES OF SOLIDS.

| Sphere, (radius, R; diameter, D)Volume | = | $\frac{4}{3}\pi$ R ³ or $\frac{1}{6}\pi$ | D3. |
|--|---|---|-----|
| ('ylinder or Prism, (height, h ; area of base, a) | - | | ah. |

Cone or Pyramid, (height, h; area of base, a) " 2/4 ah.

THE C. G. S. SYSTEM OF UNITS.

The C. G. S. System of Units is the result of an attempt to express all quantities with which physical science deals, in terms of three fundamental units;—

A Unit of Length, the centimetre:

A Unit of Mass, the gramme;

A Unit of Time, the second.

From these the following units are derived:-

Unit of Surface: the square centimetre.

- " " Volume; the cubic centimetre
- " Velocity; the velocity of one centimetre per second.
- " " Acceleration; the acceleration which imparts unit velocity to a body, in one second.
- " " Force; the dyne; the force, which, acting on a gramme mass for one second, imparts to it a unit of velocity.
- " " Work; the erg; the work done by a dyne working through one centimetre.
- " Energy; also the erg: since the energy of a body is measured by the amount of work it can do.
- " " Heat; the amount of heat required to raise one gramme of water from 0° to 1° C.
- " " Magnetic Strength; a magnetic pole has unit strength when it repels a similar pole of equal strength, one centimetre distant, with the force of a dyne.
- " " Electric Current (electro-magnetic system); a current of such strength that
 one centimetre of its circuit, bent so that every point of it is one centimetre
 distant from a unit magnetic pole, exerts upon this pole the force of a dyne.
- " ** Electric Quantity (electro-magnetic system); the quantity conveyed by a unit current in one second.
- ** Difference of Potential (electro magnetic system); two points have unit difference of potential when one erg of work must be expended to bring a unit of + electricity from one to the other against the electric force.
- " Electric Resistance (electro-magnetic system); a conductor possesses unit resistance when a unit difference of potential between its ends, causes a unit current to flow through it.

ALPHABETICAL TABLE OF UNITS USED IN PHYSICAL SCIENCE.

Ampère; unit of electric current; 10—1 C. G. S. units; the current produced by the difference of potential of a volt through the resistance of an ohm.

Calorie; French unit of heat; quantity of heat required to raise one kilogramme of water from 0° to 1° C; equals 3 908 Eng. units of heat. (See Heat, English unit of.)
Capacity, unit of electric. See Farad.

Cheval-de-vapeur. See Force-de-cheval.

Coulomb; unit of electric quantity; 10-1 C. G. S. units; quantity conveyed by the current of an ampère in a second.

Current, unit of electric. See Ampère.

Electro-motive force, unit of. See Volt.

Farad; unit of electric capacity; 10-9 C. G. S. units; quantity which, with the electromotive force of a volt, would flow through the resistance of an ohm in one second.

Foot-pound; English unit of work; work required to raise one pound through one foot in opposition to the force of gravity.

Force, units of. See Kilogramme and Pound.

Force-de cheval; French unit of power; .9864 horse-power; power of doing 75 kilogrammetres (542.5 foot-pounds) of work per second.

Heat, English unit of; heat required to raise one pound of water from 32° to 33° F. (See Calorie.)

Horse-power; English unit of power; power required to perform 550 foot-pounds of work per second.

Kilogramme: French unit of mass, and also of force. (See Pound.)

Kilogrammetre; French unit of work: work required to raise one kilogramme of mass through one metre in opposition to force of gravity.

Mass, units of. See Kilogramme and Pound.

Ohm; unit of electric resistance; 109 C. G. S. units; is the resistance offered to a current of electricity by a wire of pure silver or copper one millimetre in diameter and 48.61 metres long at 65° F (18 3° C).

Potential, unit difference of. See Volt.

Pound: English unit of mass; regarded as a weight it is used also as the unit of force, i. e. the force exercised by the mass of a pound (where g=981: London).

Power, unit of. See Force-de-cheval and Horse-power.

Power, unit of electric. See Watt.

Quantity, unit of electric. See Coulomb.

Quantity, unit of magnetic. See Weber. Resistance, unit of electric. See Ohm.

Volt; unit of electro-motive force; 10 °C. G. S. units; equals 9208 of the force of one

Watt. unit of electric power; 107 C. G. S. units; power exerted by the current of am ampère through the difference of potential of a volt.

Weber; unit of magnetic quantity; 108 C. G. S. units.

Work, units of. See Foot-pound and Kilogrammetre.

TABLE OF LOGARITHMS.

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| m | 0128 0531 0899 1239 1553 | 1847 2122 2380 2525 2855 2856 | 3075 3284 3483 3674 | 4031 4200 4362 4518 4600 | 4814 4955 5092 5224 5353 | 5478 5599 5717 5832 5944 |
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TABLE OF LOGARITHMS.

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TABLE OF ANTI-LOGARITHMS.

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TABLE OF ANTI-LOGARITHMS.

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| | Logarithm | | | Find Logarithm of 4026. Logarithm 4020 Proportional part for 6. | Logarithm required |

To Divide by use of Logarithms: Subtract the logarithm of the divisor from the logarithm of dividend; the differ-To MULTIPLY by use of Logarithms: Add together the logarithms of the numbers to be multiplied; the sum is a logarithm whose natural number is the product required.

To Extract a Root by use of Logarithms: Divide the logarithm of the number (whose root is to be extracted) ence is a logarithm whose natural number is the quotient required.

by the TO RAISE A NUMBER TO ANY POWER: Multiply the logarithm of the number (to be raised) by the index of the power; index of the root; the quotient is the logarithm of the required root. the product is the logarithm of the required number.

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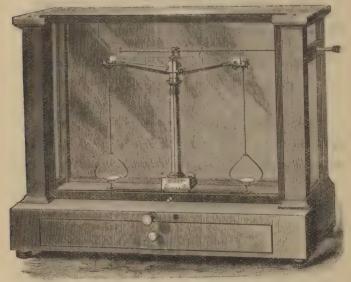
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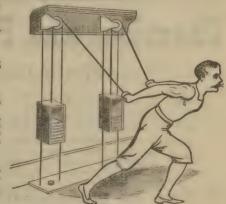
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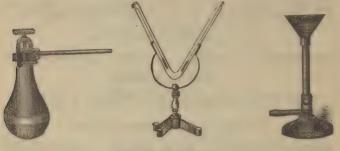
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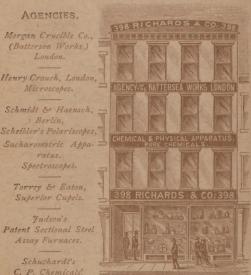
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